NMR Detection of Oxygen Isotopes in TiO₂ Single Crystal*

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We studied the electric quadrupole interactions of Oxygen isotopes in a TiO₂ single crystal. For $^{13}\mathrm{O}$ and $^{19}\mathrm{O}$ nuclei, quadrupole coupling constants were measured by the β -NMR technique, and for the $^{17}\mathrm{O}$ nucleus the FT-NMR technique was utilized. We synthesized a TiO₂ single crystal which was enriched in $^{17}\mathrm{O}$ up to 5 atom % to observe NMR signals without any perturbations from impurities. Using the known quadrupole moment of $^{17}\mathrm{O}$, EFGs at an O site in TiO₂ and the quadrupole moments of $^{13}\mathrm{O}$ and $^{19}\mathrm{O}$ were determined.

1. Introduction

β-NMR is a very useful technique for the investigation of hyperfine interactions as well as nuclear moments [1]. To observe β -NMR efficiently, it is important to select a porper sample in which the polarization of the probe nuclei is well maintained. Recently, a TiO2 single crystal was found to be a good implantation medium. For the case of ⁴¹Sc in a TiO₂ crystal, the maintained polarization reached up to 60% of the initial polarization [2], and it was reported that the TiO2 crystal preserved 100% of the polarization of ¹²N [1]. Therefore, the systematic study of the hyperfine interactions of impurities in TiO₂ is of great importance for understanding the implantation mechanism and the electronic structure of impurities. In this paper we studied the electric quadrupole interactions of oxygen isotopes in TiO₂ single crystal (tetragonal). We detected β -NMR of short-lived ¹³O ($I^{\pi} = 3/2^-$, $T_{1/2} = 8.6$ msec) and ¹⁹O ($I^{\pi} = 5/2^+$, $T_{1/2} = 27.0$ sec) in TiO₂, the crystal structure of which is of the rutile type. The parameters of the electric quadrupole coupling constants for them in TiO2 were determined.

An other aim of the present study was the determination of the electric quadrupole moment of the short-lived unstable nuclei ¹³O and ¹⁹O from the quadrupole coupling constants determined here. The knowledge of the electric field gradient (EFG) at the oxygen nucleus in the crystal is indispensable to extract the value of nuclear quadrupole moment of it. For this purpose, the Fourier-Transformed NMR (FT-NMR) was detected on the stable isotope of ¹⁷O in the crystal in a strong field at room temperature. In order to measure it accurately, the NMR signal was increased by using a newly synthesized TiO2 crystal [3] in which the enriched isotope of ¹⁷O was doped, the concentration of which was about 5% of the oxygen atoms. In the crystal no other impurities were artificially doped. The EFG at the oxygen site had been determined for a TiO2 crystal in which a certain amount of chromium was doped to make it possible to detect the NMR signal of ¹⁷O [4]. The difference of the old data from the present one is reported. These experimental values are compared with ab initio calculations of the EFG at the oxygen site in TiO₂ [5].

2. Experimental Procedure and Results

2.1 13O in TiO2

The experimental principle and techniques were essentially similar to those of the previous β -NMR measurements [1]. The ¹³O was produced through the projectile fragmentation in the reaction of ¹⁶O + Be. The primary beam of ¹⁶O with the kinetic energy of 135 MeV/u was from the Ring Cyclotron (K = 540) at RIKEN and

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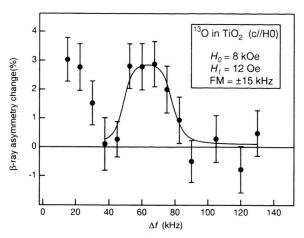


Fig. 1. β -NQR spectrum of ¹³O in TiO₂.

bombarded the 481 mg/cm² thick Be target. The ¹³O nuclei ejected at angles of 0.5° to 2.5° relative to the incident beam were separated by RIPS (RIKEN Projectile fragment Separator). As a result, the nuclear polarization was about -2.3%. Thus obtained polarized nuclei were implanted into a disc of TiO2 crystal with 0.2 mm thickness, which was placed in a magnetic field of 0.8 T. The crystal c-axis of TiO₂ was placed parallel to the external field. Under the EFG provided by the TiO2, three transition frequencies are given for ¹³O. To observe the NMR efficiently, the β -NQR method was employed, and the total polarization was completely inverted by the AFP method. For this purpose, 6 RFs, whose intensity and duration time was about 7 Oe and 1 ms, respectively, were applied. The β -NQR was detected as a function of Δf , which was defined as a half of the quadrupole splitting or the deviation from the Larmor frequency. The frequency modulation width was ±15 kHz. A typical spectrum is shown in Fig. 1. From the analysis of the spectrum, |eqQ/h| was obtained to be 1490 ± 75 kHz.

2.2 19O in TiO2

The experimental procedure was the same as that for 13 O. Production of polarized 19 O nuclei was carried out via the 18 O (d, p) 19 O reaction. An incident beam of deuterons with an energy of 3.25 MeV, extracted from the Van de Graaff accelerator at the Osaka University was used for the reaction. The deuteron beam was directed on a $\mathrm{Ti}^{18}\mathrm{O}_2$ target which was produced by oxydizing a titanium foil of 0.5 mm thickness by $^{18}\mathrm{O}$. The recoil angle of $^{19}\mathrm{O}$ was selected to be $(40 \pm 8)^\circ$. A $\mathrm{Ti}\mathrm{O}_2$ crystal was used as an implantation medium with its crystal

Table 1. Experimental and theoretical EFGs for the O site in TiO_2 . $Q(^{17}O) = 26 \pm 3$ mb is used.

	eqQ/h (kHz)	$V_{ZZ} \times 10^{15} \text{ V/cm}^2$	η
Experiment present result Gabathuler et al. [4]		240 ± 30 240 ± 30	0.831 ± 0.007 0.868 ± 0.005
Theory KKR method [5] FLAPW method [7]		-215 -196	0.94 0.84

 $\langle 1, 1, 0 \rangle$ axis set parallel to the employed external magnetic field of 0.8 T. Spin manipulations were performed by applying RF fields perpendicular to the magnetic field which was parallel to the nuclear spin, and the asymmetry change in the β -ray distribution was detected. The intensity and the duration time of the RF field was about 3.6 Oe and 30 msec, respectively. To achieve the β -NQR detection, a set of 5 RFs was applied sequentially 16 times, which was sufficient to completely depolarize the nuclear polarization. From analyses of the β -NQR results the electric quadrupole coupling constant of ¹⁹O in the TiO₂ was obtained to be 214 ± 7 kHz.

$2.3^{17}O$ in TiO_2

¹⁷O is the only stable oxygen isotope which has a finite electric quadrupole moment. Since the natural abundance of ¹⁷O is 0.04%, the NMR signal of ¹⁷O in a TiO₂ crystal is almost undetectable. Gabathuler et al. succeeded to enhance the signal of ¹⁷O by doping a certain amount of Cr, and determined the coupling constant and the asymmetry parameter to be $1497 \pm 4 \text{ kHz}$ and 0.868 ± 0.005 , respectively [4]. To get a reliable coupling constant, we synthesized a TiO2 single crystal which was enriched in ¹⁷O up to about 5 atom % and detected the FT-NMR of ¹⁷O without any perturbations from artifically doped impurities. The c-axis of the TiO₂ single crystal was set perpendicular to the external field. The transition frequencies between $m = \pm 1/2 \leftrightarrow \mp 1/2$ and $m = \pm 1/2 \leftrightarrow \pm 3/2$ were measured as a function of the rotation angle for the determination of the quadrupole coupling constant and the asymmetry parameter of the EFG. The NMR was detected under an external field of 7.0 Tesla and 9.4 Tesla for separating the quadrupole effects from the anisotropic chemical shifts. The results are shown in Figure 2. The lines are the theoretical ones best fitted to the data. The quadrupole coupling constant was obtained to be 1512 ± 4 kHz and the asymmetry parameter was determined to be 0.831 ± 0.007 .

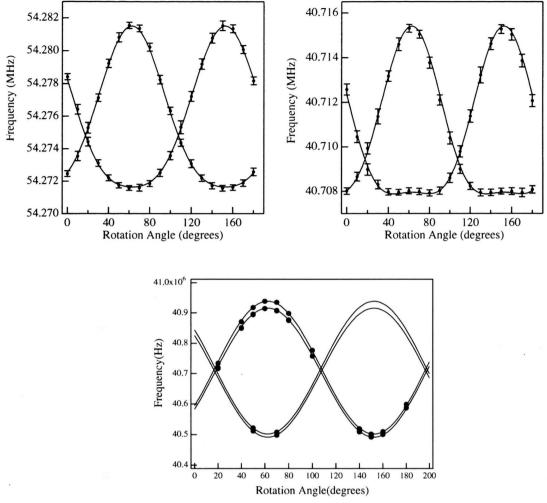


Fig. 2. The resonance frequencies for the $m = \pm 1/2 \leftrightarrow \pm 1/2$ and $m = \pm 1/2 \leftrightarrow \pm 3/2$ transitions as functions of the rotation angle.

3. Discussion

Using the known quadrupole moment of ¹⁷O [6] we extracted the EFG from the coupling constant. In Table 1, the experimental values of EFGs are summarized with Gabathuler's data. The difference of the asymmetry parameter between them suggests that a certain disturbance from the Cr might exist. The theoretical prediction by the KKR method in the framework of the local density approximation and the prediction by the FLAPW method [7] are also shown in Table 1.

Now, we obtained the EFG at the O site in TiO₂ with a high precision. Therefore, the quadrupole moment of ¹³O and ¹⁹O can be derived from the quadrupole coupling constant. Two independent sites with similar populations

of implanted nuclei were observed for 13 O as shown in Figure 1. This means that half of them was located at the substitutional site of oxygen in the crystal, which was expected to be most likely in the present implantation, and that the other half was located in a site different from the substitutional site. Assuming that the peak at the higher frequency is from the 13 O nuclei implanted at the substitutional site of oxygen, the quadrupole moment of 13 O was determined to the 26 ± 3 mb. This is comparable with the theoretical value of 16 mb by OXBASH with Cohen-Kurath wave functions. On the contrary, if we assume a substitutional location for the smaller coupling frequency of $\Delta f \sim 15$ kHz obtained from the second peak in Fig. 1, we have $|Q| \sim 6$ mb which is much smaller than the prediction. Therefore the second site cannot be sub-

stitutional. By a similar consideration, the quadrupole moment of 19 O is determined to be 3.8 \pm 0.5 mb.

In the measurements of the stable isotope, only one definite field gradient was observed, as expected. Therefore, the one extra site found in the β -NMR of ¹³O and ¹⁹O, suggests a site with shallow potential, an isometric state, produced through the implantation of the ions in the crystal.

4. Summary

We determined the quadrupole coupling constants of ¹³O, ¹⁹O and ¹⁷O in a TiO₂ single crystal. Using the known quadrupole moment of ¹⁷O, the EFG at the O site

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was extracted. The agreement with the theoretical predictions is fairly good. The quadrupole moments of ¹³O and ¹⁹O were determined from their coupling constants in TiO₂ by using the obtained EFG value. The spectrum of ¹³O and ¹⁹O suggests that there are two implantation sites in TiO₂. It is supposed that one of the extra sites is a site with shallow potential produced through the ion implantation.

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